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## A new method for the synthesis of α-thio aldehydes and alcohols from aldehydes with one-carbon elongation

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## Abstract

A two-step and high-yield method for the synthesis of  $\alpha$ -thio aldehydes from aldehydes with one-carbon elongation is realized by using chloromethyl phenyl sulfoxide as a one-carbon homologating agent. The  $\alpha$ -thio aldehydes are easily converted to desulfurized alcohols with Bu<sub>3</sub>SnH and AIBN in refluxing benzene in good yield. © 2000 Elsevier Science Ltd. All rights reserved.

The homologation of carbonyl compounds from lower carbonyl compounds by a carbon–carbon coupling reaction is an attractive way for obtaining the desired carbonyl compounds.<sup>1</sup> In addition, the produced carbonyl compounds are easily converted to other compounds useful in organic chemistry.

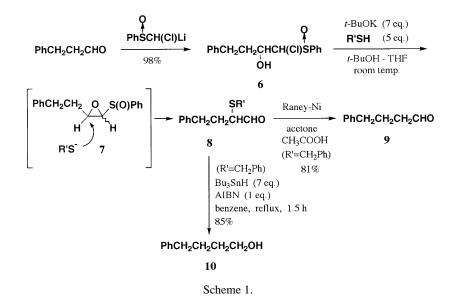
In our laboratory, we have reported some new methods for homologation of carbonyl compounds<sup>2</sup> by using aryl 1-haloalkyl sulfoxides as acyl anion equivalents.<sup>3</sup> In continuation of our work, we report herein a new method for the synthesis of  $\alpha$ -thio aldehydes **4** from aldehydes **1** with one-carbon elongation by using chloromethyl phenyl sulfoxide **2** as a one-carbon homologating agent. Moreover, the  $\alpha$ -thio aldehydes **4** were easily reduced to one-carbon elongated desulfurized alcohols **5** in good yields.

One-carbon elongation of 3-phenylpropanal is described as a specific example (Scheme 1). Treatment of the lithium carbanion of chloromethyl phenyl sulfoxide with 3-phenylpropanal at  $-78^{\circ}$ C gave the adduct **6** in quantitative yield as a diastereomeric mixture. In order to obtain the sulfinyloxirane **7**, we first treated **6** with several bases;<sup>4</sup> however, because **7** was very unstable, all the attempts gave a complex mixture. Next, we tried in situ trap of this unstable **7** in the presence of a highly nucleophilic reagent (such as thiolate, selenolate, and amine). Thus, a solution of **6** in THF was added to a mixture of *t*-BuOK (7 equiv.) and benzenethiol (5 equiv.) in a mixture of *t*-BuOH–THF (2:1) at room temperature. Quite a rapid

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reaction took place, and after 10 min the desired  $\alpha$ -(phenylthio)aldehyde 8 (R'=Ph) was obtained in 73% yield (see Table 1, entry 1). This reaction is thought to proceed via a sulfinyloxirane 7.<sup>1d</sup> We investigated this reaction with other thiols and found that alkyl thiols gave slightly better yield than benzenethiol. The best yield (93%) was obtained with benzyl mercaptan at 0°C (see Table 1, entries 2–4).



 $\alpha$ -Thio aldehydes themselves are very important compounds in organic synthesis;<sup>5</sup> however, if the  $\alpha$ -thio group can be reduced easily to hydrogen, these reactions offer a good method for homologation of aldehydes to one-carbon elongated aldehydes. To this end, reduction of the benzylthio group in **8** (R'=CH<sub>2</sub>Ph) with Raney–Ni in EtOH under several conditions was investigated; this reaction, however, only gave a complex mixture. The reduction in acetone gave some amount of **9**; however, this reaction again usually gave a complex mixture. Finally, we found that the reduction was successful in acetone containing a few drops of acetic acid to give **9** in 81% yield. However, these conditions are still not ideal. Reproducibility of this reaction was variably poor.

As the desulfurization of the  $\alpha$ -thio aldehyde was found to be rather difficult, we changed our target to a synthesis of one-carbon elongated alcohols **10**. After some investigation, reduction of **8** (R'=CH<sub>2</sub>Ph) with excess tributyltin hydride in the presence of AIBN<sup>6</sup> was found to be the conditions of choice. Reduction of **8** with 7 equiv. of tributyltin hydride with 1 equiv. of AIBN in refluxing benzene gave the desired alcohol in 85% yield (Scheme 1 and Table 1, entry 4).

Other representative examples are shown in Table 1, entries 6–12. Entries 1 and 6 show the results with *n*-alkyl aldehydes. Entries 7 and 8 show the results of the  $\alpha$ -branched alkyl aldehydes. Entries 9 to 11 show the results with aromatic aldehydes and entry 12 shows the result with  $\alpha$ , $\beta$ -unsaturated aldehydes. The yields of the adducts **3** of all the aldehydes with the lithium carbanion of chloromethyl phenyl sulfoxide are very high (87–98%). The yields for the reaction with benzyl mercaptan in the presence of *t*-BuOK gave again excellent yields of the  $\alpha$ -thio aldehydes to  $\alpha$ -thio aldehydes.<sup>7</sup>

The reduction of the  $\alpha$ -thio aldehydes 4 with Bu<sub>3</sub>SnH was examined and the results are shown in Table 1. The yields of the reduction were shown to be somewhat variable. It is important to note that because the  $\alpha$ -thio aldehyde in entry 12 was the  $\alpha$ , $\beta$ -(dibenzylthio)aldehyde the alcohol was saturated.

Table 1 One-carbon elongation of aldehydes to  $\alpha$ -thio aldehydes and alcohols

F	асно —→ RCHCн он	0 I(CI)SPh 3	R'SH i	НСНО ——	DN -	СН₂ОН 5
Ent	ry <b>RCHO</b>	3	R'SH		4	5
Ent		(Yield / %)	a) cor	ditions	(Yield / %) <sup>a)</sup>	(Yield / %) <sup>a)</sup>
1	PhCH <sub>2</sub> CH <sub>2</sub> CHO	(98)	PhSH	r.t., 10 min	(73)	
2			∕_−ѕн	r.t., 10 min	(80)	
3			CH <sub>3</sub> (CH <sub>3</sub> ) <sub>3</sub> SH	r.t., 20 min	(74)	
4			PhCH₂SH	0 °C, 20 min	(93)	(85)
5			HOOCCH₂SH	0 °C, 20 min	complex mix	
6	PhCH <sub>2</sub> O(CH <sub>2</sub> ) <sub>9</sub> CHO	(94)	PhCH₂SH	0 °C, 20 min	(93)	(98)
7	СН3	(88)	PhCH₂SH	0 °C, 20 min	(95)	(98)
8	Сно	(87)	PhCH₂SH	0 °C, 20 min	(95)	(89)
9	СНО	(87)	PhCH₂SH	0 °C, 20 min	(89)	(82)
10	СНО	(98)	PhCH₂SH	0 °C, 45 min	(98)	(79)
11	СНО	(99)	PhCH <sub>2</sub> SH	0 °C, 40 min	(98)	(91)
12	Ph	(98)	PhCH₂SH	0 °C, 20 min	(73) <sup>b)</sup>	Ph(CH <sub>2</sub> ) <sub>4</sub> OH (77)

a) Isolated yield after silica gel column chromatography.b) The product is 2,3-(dibenzylthio)-4-phenylbutanal.

We next applied the presented method to two-carbon elongation of aldehydes to  $\alpha$ ,  $\beta$ -dithio aldehydes 14 and alcohols 15. The overall process and the yields are shown in Scheme 2. All the reactions starting from 3-phenylpropanal gave over 90% yield and the overall yield was 76%. In contrast to this, the yields starting from 1-naphthaldehyde were good but not excellent, and the overall yield was reduced to 45%.

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RCI 1	PhSCH(CI)	LI 0H → RCHCHSPr CI 11	PhC	BuOK (7 eq.) C <b>H<sub>2</sub>SH</b> (5 eq.) BuOH - THF	→ RCH	H₂Ph CHO .2	Q PhSCH(CI)Li
SCH₂Ph Q I RCHCHCH(CI)SPh OH 13		<i>t</i> -BuOK (7 eq.) <b>PhCH<sub>2</sub>SH</b> (5 eq.) <i>t</i> -BuOH - THF	5 eq.) FCHCHCHO -		Bu <sub>3</sub> SnH (11 eq.) AIBN (1 eq.) benzene, reflux 1.5 - 2.5 h		RCH <sub>2</sub> CH <sub>2</sub> CH <sub>2</sub> OH
	 R	11	12	13	14	15	overall
	PhCH <sub>2</sub> CH <sub>2</sub> —	98%	93%	93%	95% <sup>a)</sup>	95%	76%
		98%	98%	79%	79%	75%	45%

a) A 2:3 diastereomeric mixture.

Scheme 2. Two-carbon elongation of aldehydes 1 to  $\alpha,\beta$ -(dibenzylthio)aldehydes 14 and alcohols 15

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